S/076/62/036/009/002/011 B101/B102

AUTHORS: Yegorov, h. M., Ignat'yeva, L. A., Kiselev, V. F., Krasil'ni-kov, h. G., and Topchiyeva, K. V.

PERIODICAL: Zhurnal fizicheskoy khimii, v. 36, no. 9, 1962, 1882 - 1889

TEXT: The specific heat of wetting of commercial Al₂O₃ by water, methanol, ethanol, and n-heptane, and the content of structural water Al₂O₃ were measured, the phase composition of Al₂O₃ was determined by x-ray analysis, and the infrared spectrum of deuterated Al₂O₃ was taken. Whereas with n-heptane the heat of wetting is independent of the content of structural water in Al₂O₃, it increases, in the case of water and alcohols, with increasing thermal dehydration of Al₂O₃. Since, however, the specific surface of Al₂O₃ becomes smaller at high annealing temperatures, the heat of Card 1/3

S/076/62/036/009/002/011 B101/B102

Study of the surface ...

wetting calculated per g of Al₂O₃ reaches a maximum for Al₂O₃ heated at 500°C. The curve for heat of wetting (Q, erg/cm³) versus structural vater (µmole/m²) shows the following sections: (1) Increase of Q after thermal treatment of Al₂O₃ at 20 - 150°C owing to removal of the adsorbed H₂O₃ (2) unchanged Q at 170 - 200°C in spite of dehydration of the bayerit in the bulk of Al₂O₃; (3) Q increases at 200 - 500°C owing to dehydration of the Al₂O₃ surfaces (4) sharp increase of Q between 500 and 700°C, although the content of structural water changes only little in this range owing to formation of γ-Al₂O₃; (5) increase of Q at 800-900°C owing to formation of K, δ, Θ, and μ-Al₂O₃ (corundum). The infrared spectrum of deuterated Al₂O₃ showed a broad 2630 cm⁻¹ band which disappeared at 400°C (interacting OD groups), a narrow band at 2755 cm⁻¹ (free, non-interacting OD groups), and a narrow 2710 cm⁻¹ band (weakly bound OD groups). For gibbsite, maximum hydration was calculated to be ~22µmole/m²; for the (0001) face of corundum, the hydration as ants to 12.7 µmole/m². The coordination sphere of the Al Card 2/3

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Study of the surface ...

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surface atoms which is not fully occupied after the thermal dehydration is filled up by water or alcohols with formation of hydrate or alcoholates, respectively. The irreversible sorption of alcohols increases after thermal treatment of Al₂O₃ at high temperature. There are 4 figures and 2

ASJOCIATION:

Moskovskiy gosudarstvennyy universitet im. M. V. Lomonosova, Fizioheskiy i khimicheskiy fakul'tety (Moscow State University imeni M. V. Lomonosov, Physical and Chemical Departments)

SUBMITTED:

November 1, 1960



Card 3/3

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s/078/62/007/002/003/019 B119/B110

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Ostroushko, Yu. I., Filippova, K. I., Ignat'yeva, L. A.

TITLE:

Card 1/3

AUTHORS:

Interaction of \$-spodumene and sulfuric acid

PERIODICAL: Zhurnal neorganicheskoy khimii, v. 7, no. 2, 1962, 244 - 251

TEXT: The mechanism of the reaction between spodumene and $\rm H_2SO_4$ was studied for varying thermal pretreatment of the former. β -spodumene was obtained from α -spodumene ($\rm Li_2O$ - 6.71%; $\rm Al_2O_3$ - 23.94%; $\rm SiO_2$ - 62.4%) by heating to 1000°C (tube Silit furnace). The conversion of the α to the β -form was checked by crystal optical and x-ray analyses. β -spodumene was made to react with $\rm H_2SO_4$ in quartz test tubes (standard conditions: 250°C for 60 min; $\rm H_2SO_4$ consumption 40%), the mixture was filtered and washed with hot water. Residues were studied by x-ray diffraction analysis (with the $\rm VP(-70$ (URS-70) apparatus), with the $\rm PKA-62$ (RKD-62) camera with Fe anode and Mn filter) as well as infrared spectrographically (NKC-2 (IKS-2) double-beam infrared spectrograph with LiF prism for the range from 6000

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Interaction of \$-spodumene...

Card 2/3

to 1500 $\,\mathrm{cm}^{-1}$ and with KCl prism from 1400 to 550 $\,\mathrm{cm}^{-1}$) and compared with the data of the initial substances. Besides these analytic methods thermogravimetric and chemical analyses were used. 1) The minimum tempering temperature for α-spodumene required for a reaction with H₂SO₄ (it is 950°C), 2) the optimum temperature and time of the spodumene - H2SOA reaction (up to 100°C - spodumene is not changed; minimum reaction temperature 150°C, optimum temperature with minimum reaction time 250 - 300°C); 3) the reversibility of the reaction with H2SO4 by tempering of the non-washed reaction product at 500, 700, 800, 900, 1000, and 1100°C were determined. Results: β -spodumene reacts with H_2SO_4 as follows: $\text{Li}_20\cdot\text{Al}_20_3\cdot\text{4}$ Si0_2 + $\text{H}_2\text{S0}_4$ \longrightarrow $\text{Li}_2\text{S0}_4$ + $\text{H}_20\cdot\text{Al}_20_3\cdot\text{4}$ Si0_2 ; the IR spectrogram of the residue shows one OH-vibrational band each at 3020 and 2450 cm-1 (the latter verified by substituting H20 by D20) which are not present in the spectrogram of the initial substance. The above-mentioned reaction is not possible with α-spodumene. Significant deformations of the crystal lattice occur, if Li in spodumene is replaced by H. The residue resulting

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Interaction of \$-spodumene...

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after leaching is a particular mineral which is not like the product leached under natural conditions. The substitution reaction effected by is reversible above 700°C: \$\beta\$-spodumene is formed again. There are four references to English-language publications read as follows:

I. J. Bear. Chem. Engng. and Mining Rev., 50, 40 (Febr. 1958); I. J. Bear. Chem. and Engng. News, 32, no. 29, 2868; no. 51, 5017; no. 52, 5108 (1954); Lo.E. Djigheuzian. Symposium on the extraction metallurgy of some of the Metallurgical Developments in the Recovery of Some of the less Common Metals in Canada; R. Hader, R. Nielsen, M. Herre. Ind. Engng. Chem., 43

SUBMITTED: February 20, 1961

X

Card 3/3

IGNAT'YEVA, L.A.; LEVSHIN, L.V.; OSIPOVA, T.D.; POLUKHIN, Yu.M.

Study of the association of rhodamine 6G molecules based on electron and vibrational absorption spectra. Opt. i spektrents no.3:396-402 S '62. (Mira 15:9) (Mira 15:9)

Discussion of V.L.Levshin's report "Migration of energy in solutions and the association theory of the quenching of luminescence." Izv. AN SSSR. Ser. fiz. 26 no.1:52 Jz. 162. (Solution(Chemistry)) (Luminescence) (Levshin, V.L.)

APPROVED FOR RELEASE: 04/03/2001 CIA-RDP86-00513R000518410014-7"

APPROVED FOR RELEASE: 04/03/2001

CIA-RDP86-00513R000518410014-7"

BORISOVA, M.S.; DZIS'KO, V.A.; IGNAT'YEVA, L.A.; TIMOFEYEVA, L.N.

(Spectrum, Infrared)

Acidity of hydroxyl groups of oxide catalyst surfaces studied by means of infrared spectroscopy. Kin. i kat. 4 no.3: 461-466 My-Je 163. (MIRA 16:7)

1. Moskovskiy gosudarstvennyy universitet imeni Lomonosova, fizicheskiy fakulitet i Fiziko-khimicheskiy institut imeni Karpova.

(Catalyats) (Hydroxyl group)

YEGOROV, M.M.; IGNAT'YEVA, L.A.; KISELEV, V.F.; KRASIL'NIKOV, K.G.;

Surface properties of catalytically active aluminum oxide. Zhur. fiz. khim. 36 no.9:1882-1889 S '62. (MIRA 17:6)

l. Moskovskiy gosudarstvennyy universitet imeni Lomonosova, fizicheskiy fakul'tet i khimicheskiy fakul'tet.

IGNATIVEVA, L.A.; SIDERITO, A.Ta.; SECVENISTONA, T.A.

Infrared spectroscopy study of the transformations of isomeris cresols on NI/Al203-catalysts. Kin.i kat. 5 no.621069-1075 N-D (MIRA 18:3)

1. Moskovskiy gosudarstvennyy universitet imeni Lomenosova, fizicheskiy i khimicheskiy fakulitety.

IGNAT'YEVA, L.A.; MUSAYEV, T.N.; SLOVOKHOTOVA, T.A.

Study of interaction of isopropyl alcohol with a Ni/Al₂O₃ catalyst by infrared spectroscopy. Kin. 1 kat. 6 no.2:294-299 Mr-Ap *65.

1. Moskovskiy gosudarstvennyy universitet imeni Lomonosova, fizicheskiy i khimicheskiy fakul'tety.

Role of hydroxyl groups at the surface of exide catalysts in adsorption and catalysis processes. Dokl. AN SSSR 163 no.2:398-401 Jl '65. 1. Moskovskiy gosudarstvennyy universitet. Submitted January 4, 1965.

TAYMURAZOVA, L.Kh.; IGNAT'YEVA, L.A.

Study of the interaction between minerals and polymers by infrared spectroscopy wethod. Vest. Mosk. un. Ser. 6: Biol., pochv. 20 no.2: 81-86 Mr-Ap 165. (MIRA 18:5)

1. Kafedra fiziki i melicratsii pochv Moskovskogo universiteta.

DAVYDOVA, N.I.; ZHIGUNOVA, I.A.; IGNAT'YEVA, L.A.; KOVNER, M.A.

Commence of the commence of th

Calculation and interpretation of the spectra of nonplanar vibrations in m-cresol, n-cresol, o-cresol and their deuterosubstituted. Opt. i spektr. 18 no.6:1077-1079 Je 165.

(MIRA 18:12)

IGNAT'YEVA, L.A.

Determining the productivity c" the serial part of grasses in a birch and aspen forest. Izv. SO AN SSSR no.8.Ser.biol.-med.nauk no.2:62-67 '65. (MIRA 18:9)

1. TSentral'nyy Sibirskiy botanicheskiy sad Sibirskogo otdeleniya AN SSSR, Novosibirsk.

IGNATIYEVA, L.A.; TUMAHOVA, L.A.; AKIMOVA, M.V.

Studying the effect of a catalytic poison on the hydroxy' coating of oxidic catalysts by the infrared spectroscopy method. Zhur.prikl. spekt. 2 no.4:331-335 Ap *65.

(MIRA 18:8)

APPROVED FOR RELEASE: 04/03/2001 CIA-RDP86-00513R000518410014-7"

IGNAT'YEVA, L.G.

Development and distribution of irrigation farming in the Crimean steppes. Izv. Krym. otd. Geog. ob-va no.5:233-243 '58.

(Grimea--Irrigation farming)

MEZENTSEV, Mikhail Denilovich; CHETYRKIM, M.I., otvetstvennyy redektor;
SUROVA, V.A., redektor izdatel'stve; IOMAT'TEVA, L.I., redektor
izdatel'stva; ZAZUL'SERIA, V.F., tekhnicmsky redektor;
KOROVEHKOVA, Z.A., tekhnicheskiy redektor

[The economics, organization and planning of production in the
ocal industry] Ekonomika, organizatiia i planirovanie proisvodstva v ugol'and promyshlennosti. Isd. 2-oe, perer. i dop. Moskva,
Ugletekhizdat, 1956. 342 p.

(Goal mines and mining)

BUKHALO, Sergey Makeimovich; GERCHIKOV, S.S., otvetstvennyy radaktor; SUROVA, V.A., redaktor isdatel stva; IGNAT: TRVA, L.I., redaktor izdatel'stva; ALADOVA, Ye.I., tekhnicheskiy redaktor [Organization and planning of production in coal mines] Organizatsiis i planirovanie proisvodstva na ugolinykh shakhtakh. Moskva, (MIRA 10:8) Ugletekhisdat, 1957. 355 p. (Coel mines and mining)

> CIA-RDP86-00513R000518410014-7" APPROVED FOR RELEASE: 04/03/2001

RZHEVSKY, Vladimir Vasil'yevich; SIMKIN, B.A., otvetstvennyy red.;
SUROVA, V.A., red.; IGMAT'IEVA, L.I., red.; EKKKER, O.G., tekhn.red.

[Open-cut mining of cosl and ore] Rezhim gornykh rabot pri otkrytoi dobyche uglia i rudy. [Moskva] Ugletekhizdat, 1957. 198 p.

(NIRA 11:1)

(Strip mining)

SKOCHINSKIY, A.A., akademik, red.; TERPIGOREV, A.M., akademik; SHEVYAKOV, L.D., akademik, red.; MEL'NIKOV, N.V., red.; AGOSHKOV, M.I., red.; SPIVAKOVSKIY, A.O., red.; PLAKSIN. I.N., red.; SUDOPLATOV, A.P.; doktor tekhn.nauk; red.; BARON, L.I., doktor tekhn.nauk, red.; PROTOD'YAKONOV, M.M., doktor tekhn.nauk, red.; TAYERMAN, Ye.M., doktor tekhn.nauk, red.; TAYERMAN, Ye.M., IGNAT'YEVA, L.I., red.; BEKKER, O.G., tekhn.red.; ALADOVA, Ye.I., tekhn.red.

[Soviet mine engineering, 1917-1957] Sovetskaia gornaia nauka, 1917-1957. Moskva, Gos.nauchno-tekhn.izd-vo lit-ry po ugol'noi promyshlennosi "Ugletekhizdat," 1957. 640 p. (MIRA 11:1)

1. Akademiya nauk SSSR, Institut gornovo dela. 2. Chlen-korrespondent AN SSSR (for Mel'nikov, Agoshkov, Spivakovskiy, Plaksin).

(Mining engineering)

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CIA-RDP86-00513R000518410014-7

IJP(c) EWT(m)/T/EWP(t)/ETI L 36113-66 SOUNCE CODE: UR/0126/66/021/005/0700/0793 ACC NR: A. 6017304 KUTHORS: Palatnik, L. S.; Ignat'yev, O. H.; Ignat'yeva, L. K. B UKG: wharkov Polytechnic Institute im. V. I. Lenin (Khar'kovskiy Politekhnicheskiy institut); Institute of Chemistry and Technology of Mare Elements, Kol'sk Branch AN SSJR (Institut khimii i tekhnologii redkikh elementov Kol'skogo filiala AN SSSR) TITLE: Method of curvilinear supports for the proparation of complete alloy systems of variable composition after the method of S. A. Vokshinskiy SOURCE: Fizika metallov i metallovedeniye, v. 21, no. 5, 1966, 700-703 TOPIC TAGS: alloy, alloy composition, alloy phase diagram, alloy system, metal vapor deposition ABSTRACT: A mothod for the simultaneous preparation of two- and three-component alloy systems covering the complete concentration range of all components is presented The new method is an extension of the one propos d by S. A. Vekshinskiy (Novyy metod metallograficheskogo issledovaniya splavov, N., Gostokhizdat, 1944). The method consists of a simultaneous vacuum evaporation of all the alloy components onto a spherical or cylindrical surface (see Fig. 1). The density of condensate at a given point (see Fig. 1) is given by the expression $Q[(b+1)\cos a-b]$ $4\pi R^{2} [2b(b+1)(1-\cos a)+a^{2}+1]^{6/6}$ 539.216.2 UDC: Card 1/2

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CIA-RDP86-00513R000518410014-7

Fig. 1. a - condensation onto the outer surface of a cylinder from a point source evaporator; b - production of a binary condensate on a cylindrical or spherical condenser;

D A and A' - evaporated components;

EB' - region of condensation of the binary alloy of variable composition; ED - region of condensation of pure component A; B'D' - region of condensation of condensation of condensation of pure component

where Q is the mass of the evaporated substance, R is the distance between the evaporator and the epicenter, b = r/R is a geometrical factor, $a = C_X/R$ is the linear coordinate of point C. This relationship was tested experimentally on antimony specimens, and good agreement between the calculated and experimental values for q was obtained. A photograph of the experimental apparatus is presented. Orig. art. has: 5 figures and 2 equations.

SUB CODE: 11/

SUBM DATE: 12Jun65/

ORIG REF: Oll

LS

Card 2/2

SOURCE: Leningrad. Universitet. Vestnik. Seriya matematiki, mekhaniki i astronomii, no. 1, 1965, 102-109

TOPIC TAGS: solar eclipse, solar atmosphere, residual radiation, terrestrial atmosphere, radio emission, sunspot

ABSTRACT. An expedition went to Simushir Island to observe the time of the second and third radio contacts of the solar eclipse of 21 July 1963 for detecting the neight of rapid changes in the solar atmosphere during the period of weak solar action of the for measuring the residual radiation that the form the period of total and the form. The detection of the action of the solar actions are an interesting assertions.

ABBASOV, A.R.; GREBIESKIY, A.S.; DERAGOVA, M.S.; DUPPOV, V.A.; ISMATYEWA, L.M.; MOLCHAHOV, A.P.; HYASHIKOV, V.L.; PARKRATOV, Ye.L.; UTKHAHOV, A.G.; YUDIH, O.I.; YASHOV, L.V.

Radionstronomical observations of the solar eclipse of July 21, 1963 in the microwave band. Vest. LGU 20 no.1:102-109 '65. (MIRA 18:2)

APPROVED FOR RELEASE: 04/03/2001 CIA-RDP86-00513R000518410014-7"



SOLINOV, F.G., kand.tekhn.nauk; BUDOV, V.M., inzh.; KRUCHININ, Yu.D., kand. tekhn.nauk; IGNAT'TEVA, L.M., inzh.

Effect of additions of fluorine and the replacement of sodium oxide by potassium oxide on the crystallizing properties of sheet glass. Stek. i ker. 22 no.6:22-25 Je 165.

1. Gosudarstvennyy nauchno-issledovateliskiy institut stekla (for Solinov). 2. Salavatskiy zavod tekhnicheskogo stekla (for Budov). 3. Uraliskiy politekhnicheskiy institut imeni S.M. Kirova (for Kruchinin, Ignatiyeva).

APPROVED FOR RELEASE: 04/03/2001 CIA-RDP86-00513R000518410014-7

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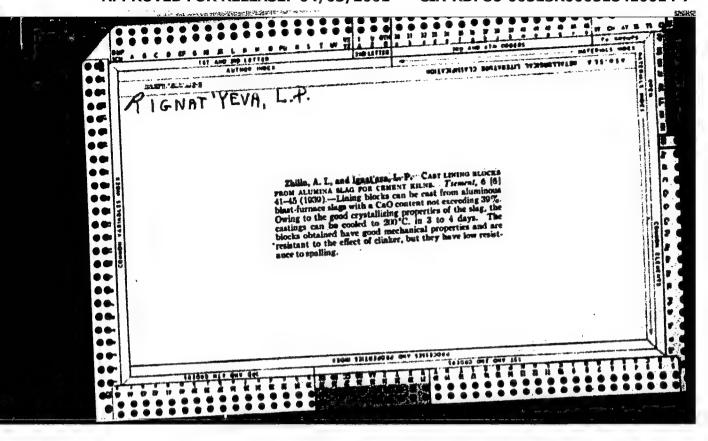
IGNATIYEVA. L.N., inzh.

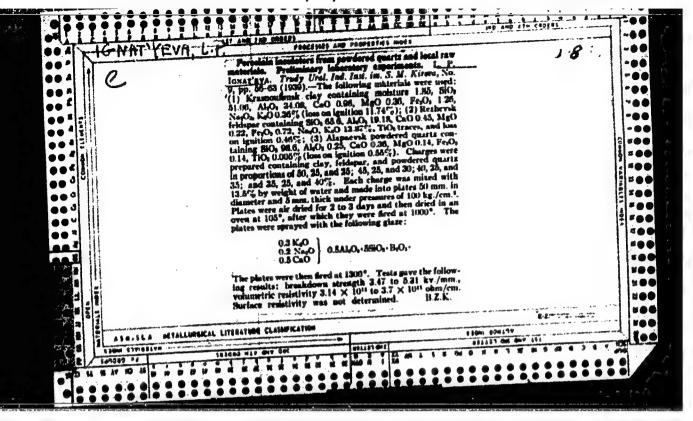
Reconstruction of the air ducts of the ejector-type dryer designed by the Central Scientific Research Institute of Woodworking. Der. (MIRA 17:4) prom. 13 no.4:22-23 Ap '64.

l. Tiraspol'skaya mebel'naya fabrika No.4.

"APPROVED FOR RELEASE: 04/03/2001

CIA-RDP86-00513R000518410014-7





CHEBUKOV, M. F., kand. tekhn. nauk; IGMAT! YEVA, L. P., insh.

Building gypsum made of wastes obtained in producing hydrofluoric acid. Stroi. mat. 6 no:10:36 0 *60. (MIRA 13:10) (Gypsum)

ार्ड १६ - १९१५ मार्ड्स्स वर्षे केंग्रेज्य संस्थातम् सार्धाम्यामस्य संस्थातस्य

CHEBUKOV, M.F.; IGNAT'YEVA, L.P.

Hydrofluoric acid production wastes as additives to cement for regulating the time of setting. Zhur. VKHO 5 no.6:712-713 '60. (MIRA 13:12)

1. Ural'skiy politekhnicheskiy institut im. S.M.Kirova. (Cement) (Hydrofluoric acid)

IGNAT'YEVA, L.P. Dessication of cement slurries by centrifugation. Trudy Ural. politekh. inst. no.118:32-37 '62. (MIRA 16:6) (Gement) (Separators(Machines))

IGNAT'YEVA, L.P.

Neutralized gypsum wastes from the production of hydrefluoric acid as an additive during clinker grinding. Trudy Ural. politekh. inst. no.118:38-43 '62. (MIRA 16:6)

(Cement-Testing) (Cypsum)

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ACCESSION NR: APh007983

\$/0190/63/005/012/1850/1853

AUTHORS: Razinskaya, I. N.; Kozlov, P. V.; Shtarkman, B. P.; Ignat'yeva, L. P.

TITLE: Intra- and interbundle plasticization of poly(vinyl chloride) interbundle

SOURCE: Vy*sokomolekulyarny*ye soyedineniya, v. 5, no. 12, 1963, 1850-1853

TOPIC TAGS: polymer, poly(vinyl chloride), polymerization, emulsion polymerization, bulk polymerization, plasticization, intrabundle plasticization, interbundle plasticization, mixed plasticization, plasticizer, primary supermolecular structure, supermolecular structure, secondary structure, bundle, glass transition temperature, PVC

ABSTRACT: The plasticization of polyvinylchloride (PVC) propared by suspension polymerization (PF-4) and block polymerization has been investigated. The compounds used as plasticizers were: dioctylphthalate, ethylstearate, butylstearate, castor oil, and glycerine. The investigation was carried out by the thermomechanical method with specimens prepared from pressed powders. Three types of plasticization are shown for PVC: intrabundle, interbundle, and a combination of these two limiting types. Because of the greater effect of plasticization of PF-4 than

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ACCESSION NR: APLO07983

of the block polymer it has been suggested that the former is endowed with looser primary supermolecular structures. It has been shown that the plasticization effect is not changed qualitatively or quantitatively in all three types of plasticization on reprecipitation of PVC from dilute solution. This is ascribed to retention of the primary supermolecular structures (bundles) during this process. Orig. art. has: 1 figure.

ASSOCIATION: none

SUBMITTED: 22Jun62

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DATE ACQ: 20Jan64

ENCL: 00

SUB CODE: MA

'NO REF SOV: 006

OTHER: 000

Card 2/2

ACC NRI AP	014 0101/00/001/0012/0013	•
AUTHOR: (Chebukov, M. F. (Professor); Ignat'yeva, L. P. (Candidate of sciences)	
ORG: Ural	Polytechnic Institute (Ural'skiy politekhnicheskiy institut)	:
TITLE: Bo	oric acid from ores	
SOURCE: 7	rsement, no. 1, 1967, 12-13	į į
TOPIC TAGS	ic seil, leaste boron mineral, datolite, gypsum,	
	The Urals Scientific Research Chemical Institute has developed a method for obtaining boric acid from datolite and lean borate ores from Far Eastern regions. The method is based on grinding rocks and leaching them with sulfuric acid. Large amounts of gypsum are obtained as a by-product. It is suggested that gypsum-rich by-products of the datolite processing be used at the Far Eastern cement plants as additives to clinkers instead of gypsum imported from the central regions of the	
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ACC NR. AP7007510

USSR. These Far Eastern cement plants can consume up to 100,000 ton of gypsum yearly [the information is of interest because it indicates the potential scale of datolite rock processed for boron compounds]. Orig. art. has: 1 table. [NC]

SUB CODE: 07 04 SUBM DATE: none/ ATD PRESS: 5117

Card 2/2

1940。基本的基本 经编数程序

A TO THE OWNER OF THE PROPERTY OF THE PROPERTY

IGNAT'YEVA, L.V.

Self-made polaroid-ocular photometer. Per. zvezdy 14 no.2: 119-121 Je 162. (MIRA 17:2)

l. Astronomicheskaya observatoriya Moskovskogo gosudarstvennogo pedagogicheskogo instituta imeni V.I. Lenina.

IGNAT'YEVA, M. A.

"The Pharmacological Characteristics of a New Domestic Alkaloid, Triakantin (triacanthine)." Cand Med Sci, Leningrad Sanitary-Hygiene Medical Inst, Min Health RSFSR, Leningrad, 1955. (KL, No 13, Mar 55)

SO: Sum. No. 670, 29 Sep 55--Survey of Scientific and Technical Dissertations Defended at USSR Higher Educational Institutions (15)

1.2. 政策制度的编码。

TOMILINA, T.N.; FOSKALFIKO, A.N.; MALYGINA, Ye.1.; IGLA. YEVA, M.A.; ANICHKOV, S.V., prof., red.; FYEETINA, A.A., red.

[Practical work in pharmacology] Fraktikum po farnakologiii. Moskva, Meditsina, 1965. 189 p. (MIRA 18:2)

1. Deystvitelinyy chlon AMN SSSR (for Anichkov).

USSR/Pharmacology. Toxicology. Cardiovascular Drugs

Abs Jour : Ref Zhur - Biol., No II, 1958, No 52005

Author | Ignat'yeva M.A.

Inst :

Title

: Hypotensive and Spasmolytic Effects of Trincantine.

Orig Pub : Farmakol. 1 toksikologiya, 1957, 20, No 1, 56-58

Abstract: Triacantine (I) is an alkaloid of Gleditychia triacanta L.

It was established that DL₅₀ of I for mice is 259 mg/kg. In intravenous administration of I to decerebrated cat, in doses of 0.1-5 mg/kg, a hypotensive effect (HE) was noted, increasing with larger doses; the HE of I is 10 times weaker than that of papaverine. Preliminary section of the vagus nerve or injection of atropine had no effect of the HE of I. Denervation of the carotid sinuses was also without effect on the HE of I. The establishment of HE in animals with damage of the spinal cord proved a direct effect of I on the cardio-vascular system. This was confirmed by experiments with an isolated heart and blood

Card: 1/2 Chair of Pharmocology, Leningrad Sanitzy Hygine Mes. Inch.

Iffect of certain marcotics on the periodic activity of an empty stomach in a dog. Farm.i toks. 22 no.5:395-397 S-0 '59. (MIRA 13:3) 1. Kafedra farmakologii (saveduyushchiy - prof. S.V. Anichkov) Leningradskogo sanitarno-gigiyenicheskogo meditsinskogc instituta. (MARCOTICS pharmacol.) (STOMACH pharmacol.)

APPROVED FOR RELEASE: 04/03/2001 CIA-RDP86-00513R000518410014-7"

1 - 3 :.

IGNATIVEVA Matrone Alaksandrovna; POLYAK, O.B., red.; RODIONOVA, Z.A., red.; KREYS, I.G., tekhn. red.

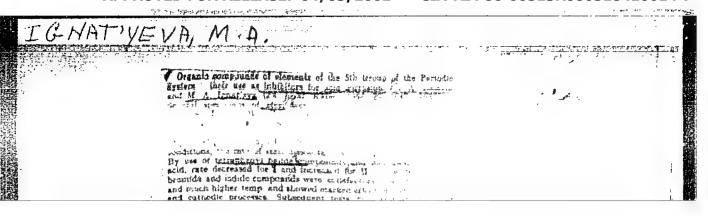
[Developing independent solution of problems in the first grade]
Privitie navykov samostoiatellnogo resheniia madach v I classe.
Pod red. G.B. Poliaka. Moskva, Gos. uchebno-pedagog. izd-vo M-va
prosv. RSFSR, 1957. 69 p.

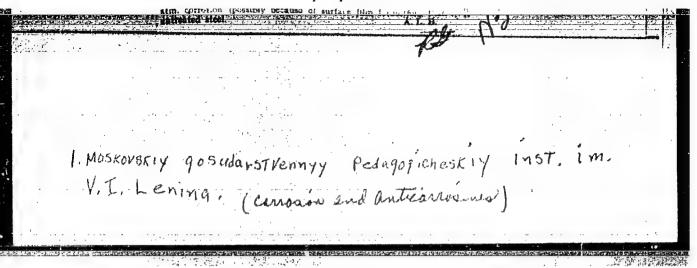
(Arithmetic-Study and teaching)

IGNAT'YEVA, H. A.

IGNAT'YEVA, M. A. - "Effect of Organoelemental Compounds of the Fifth Group of the Periodic System of Elements of D. I. Mendeleyev on the Velocity of Solution of Steel in Inorganic Acids." Sub 3 Mar 52, Moscow State Pedagogical Inst imeni V. I. Lenin. (Dissertation for the Degree of Candidate in Chemical Sciences).

50: Vechernaya Moskva January-December 1952

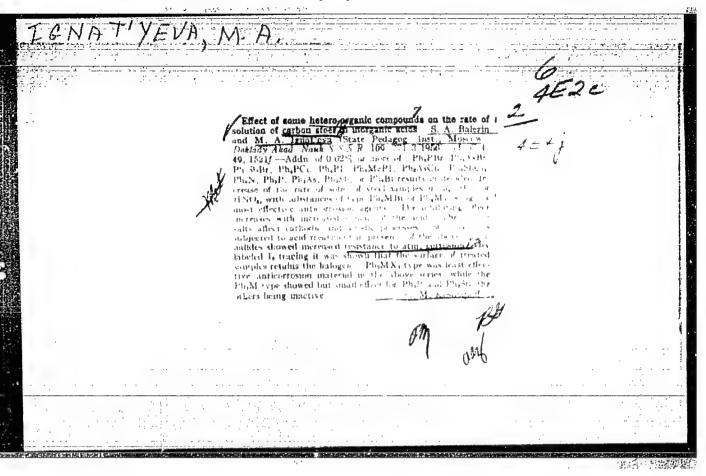




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Effer. A some heterologonic compounds on the rate of solution of carbon steel in horganic acids. S. A. Balezin and M. A. Ignat eva. Proc. Acid. Sci. U.S.S.R., Sect. Chem. 199, 489-01(1999)(Baglish translation). Sec C.A. St. 21, 3413f. 2 1 4 5 2.	
Effect: At some heterologyanic compounds on the rate of solution of carbon steel in inorganic acids. S. A. Balezin and M. A. Ignation. Proc. Acid. Sci. U.S.S.K., Sect. Christ. 199, 400 (1959) (English translation).—Sec C.A. S. L. S. R. S. L. S. R. S. L. S. L	
Effect: At some heterologyanic compounds on the rate of solution of carbon steel in inorganic acids. S. A. Balezin and M. A. Ignation. Proc. Acid. Sci. U.S.S.K., Sect. Christ. 199, 400 (1959) (English translation).—Sec C.A. S. L. S. R. S. L. S. R. S. L. S. L	
Effect: A some helpsoliganic compounds on the rate of solution of carbon steel in horganic acids. S. A. Balezin and M. A. Ignaticva. Proc. Acid. Sci. U.S.S.K., Sect. Chem. 199, 489-91(1999)(English translation).—Sec C.A. Sci. 3413f.	
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Effer, M some heterologanic compounds on the rate of solution of carbon steel in inorganic acids. S. A. Bulezin and M. A. Ignat eva. Proc. Acid. Sci. U.S.S.R., Sect. 1 Chem. 109, 423 01(1959) (Raglish translation).—Sec C.A. 1 St. 3413f. 2	
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Effect, it some helacoffequate compounds on the rate of solution of carbon steel in horganic acids. S. A. Balezin and M. A. Iguat'eva. Proc. Acid. Sci. U.S.S.R., Sect. 1 Chem. 100, 450 01(1000)(Reglish translation).—See C.A. 1 St. 2413f. 2 4 f. 2 c. 4 f. 4 f.	
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SOV/137-58-10-21322

Translation from: Referativnyy zhurnal, Metallurgiya, 1958, Nr 10, p 125 (USSR)

AUTHORS: Balezin, S. A., Ignat'yeva, M. A.

TITLE: Influence of Organogen Compounds on the Rate of Dissolution of

Steel in Mineral Acids (Vliyaniye elementorganicheskikh soyedineniy na skorost' rastvoreniya stali v neorganicheskikh

kislotakh)

PERIODICAL: Uch. zap. Mosk. gos. ped. in-ta, 1957, Vol 99, pp 77-86

ABSTRACT: A study of the influence of tetraphenyl bromides of elements of the fifth group: [(C₆H₅)₄PBr, (C₆H₅)₄AsBr, (C₆H₅)₄SbBr];

organic compounds containing phosphorus: [(C₆H₅)₄PC₁, (C₆H₅)₄PI, (C₆H₅)₃CH₃PI]; diphenyl chlorides of As and Sb: [(C₆H₅)₂AsC₁₂ and (C₆H₅)₂SbC₁₂]; and thriphenyl compounds containing N, P, Sb, Bi, and [(C₆H₅)₃N, (C₆H₅)₃P, (C₆H₅)₃As, (C₆H₅)₃Sb, (C₆H₅)₃Bi], on the rate of dissolution of steel in

H₂SO₄ (1 - 10N) and HC1 (1 to 5N) solutions at 25°C within 3 - 6

Card 1/2 hours. It is shown that the greatest inhibition of the rate of

SOV/137-58-10-21322

Influence of Organogen Compounds on the Rate of Dissolution (cont.)

dissolution of steel in $\rm H_2SO_4$ takes place with a concentration of the inhibitor as low as 0.5 millimole/1 of solution. Upon a further increase in the concentration of the inhibitor the rate of dissolution is almost unchanged. With HCl the inhibiting effect increases without interruption with an increase in the concentration of the inhibitor. In $\rm H_2SO_4$ inhibited by tetraphenyl halogenides, the rate of dissolution decreases with an increase in the concentration of acid; in HCl the rate increases with the increase in the concentration of the acid. Tetraphenyl bromides and iodides cause a considerable retardation of the process of dissolution in the 25 - 60° temperature range. It is shown that tetraphenyl halogenides affect the rates of anodic and cathodic processes. Tetraphenyl compounds proved to be stronger inhibitors than diphenyl trichlorides. Triphenyl compounds inhibit the dissolution of steel in $\rm H_2SO_4$ to a still smaller degree.

1. Steel--Decomposition 2. Acids--Chemical reactions 3. Organic com-- L. A. pounds--Chemical effects 4. Metal bromides--Chemical effects 5. Metal chlorides -- Chemical effects

Card 2/2

EWT(1)/EWT(m)/EWP(t)/EWP(b) LJP(c) UR/0058/65/000/004/D032/D032 ACCESSION NR: ARSO14397 SOURCE: Ref. zh. Fizika, Abs. 4D242 I,; Melik-Gaykazyan, I. Ya.; Grigoruk, L. V.

TITLE: Effect of lead impurity on the concentration of F-centers in alkali halide

phosphor crystals

CITED SOURCE: Sb. Spektroskopiya. M., Nauka, 1964, 176-178

TOPIC TAGS: crystal phosphor, color center, alkali halide, sodium chloride, potassium chloride, potassium bromide

TRANSLATION: The authors study the effect of Pb-content on the number of F-centers (n_p) in NaCl-Pb, KCl-Pb and KBr-Pb crystal phosphors. The Pb-content (C_{\max}) is determined which corresponds to the maximum number of F-centers. The initial growth in np as the activator concentration is increased is due to embedding of the impurity into the fundamental lattice structure at concentrations less than C max which increases the concentration of V- and then F-centers. The reduction in Fband absorption with a further increase in Pb-content is associated with that por-

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L 9671-66 EWT(1)/T - IJP(c)

SOURCE CODE:

UR/0181/65/007/011/3465/3467

29

ACC NR: AP5027452

94.55 44.55 AUTHOR: Melik-Gaykazyan, I. Ya.; Roshchina, L. I.; Ignat'yeva, H. I.

14,55

B

ORG: Tomsk Polytechnical Institute im. S. M. Kirov (Tomskiy politekhnicheskiy institut)

TITLE: Accumulation of P-centers in KCl crystals with an admixture of sulfur

SOURCE: Fizika tverdogo tela, v. 7, no. 11, 1965, 3465-3467

21.44.55 TOPIC TAGS: sulfur, potassium chloride, crystal defect, color center

ABSTRACT: The number of anion vacancies in a KC1 crystal was increased by adding 1 mol \$ Na2S to the melt, thus reducing the concentration of cation vacancies. The state of the cation sublattice with respect to defects was checked by measuring the electrical conductivity in the low-temperature region. Curves for conductivity as a function of temperature show that the conductivity of the doped crystal is two orders of magnitude lower at 120°C than that of the pure KC1 crystal at the same temperature. This indicates a reduction in the concentration of isolated cation vacancies, which causes a reduction in the rate at which P-centers are generated on preradiation defects in a KCl·S crystal in comparison with pure KCl. Experimental data are given for the rate of accumulation of F-centers on vacancies produced by radiation, as well as for other parameters of F-center kinetics in both doped and pure KC1. It was found

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EWT(m)/T/EWP(t)/ETI IJP(c) L 36394-66 RM/JD/JG SOURCE CODE: UR/0070/66/011/003/0410/0414 ACC NR. AP6018769 AUTHOR: Melik-Gaykazyan, I. Ya.; Ignat'yeva, M. I. ORG: Tomsk Polytechnical Institute (Tomskiy politekhnicheskiy institut) TITLE: Thermal and radiative dissociation of complexes in alkali-halide crystals alloyed with divalent additions SOURCE: Kristallografiya, v. 11, no. 3, 1966, 410-414 TOPIC TAGS: alkali halide, impurity content, impurity conductivity, cation, defect structure, thermal conductivity, thermal decomposition, x ray irradiation ABSTRACT: The dissociation of metal-vacancy complexes (M++v+) by heat and x-ray irradiation was studied in the alkali-halide crystals: NaCl-Mn++, NaCl-Cd++. KCl-Pb++. KCl-Sr ++ and KBr-Pb++. Electroconductivity, microhardness and the density of color centers were measured as a function of impurity content. The electrical conductivity was measured as a function of temperature (30° to 380°C) for impurity contents up to 1.3 at %; the density of F-centers were determined from the absorption coefficients in the maximum F-region using spectrophotometer readings; the concentrations of Mn*+ Cd and Sr were determined by colorimetric titration. A constant irradiation dose UDC : 548 Card 1/3

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ACC NR: AP6018769

of about 10^4 roentgens was used. The temperature dependence for \log (σ) (conductivity) was linear and an activation energy of 0.4 ev was calculated for Mn v complexes in the temperature range 200-500°C. The dissociation of complexes increased the number of single cation vacancies in the lattice and increased the conductivity above 200°C. In contrast to NaCl-Cd⁺⁺ and NaCl-Mn⁺⁺ the conductivity of KCl-Pb⁺⁺, KCl-Sr⁺⁺ and KBr-Pb++ increased with crystal purity for temperatures of 20-200°C. The dependence of log (a) on impurity concentration was given for crystals in different conditions. In all cases, the curve rose sharply and leveled out at concentrations lower than the limit of solid solubility for the particular systems. The microhardness, indicating the degree of resistance to plastic deformation, was highly dependent on the introduction of divalent ions into NaCl. At temperatures corresponding to complete dissociation of complexes (indicated by electroconductivity) the abscissa dropped for log $(\sigma)=f(\sigma)$ and H (microhardness) = $f'(\sigma)$. Further increases in temperature did not affect the lebel of the curves. Irradiation dropped the conductivity as a result of the increase in the concentration of electron-acceptor impurities (Pb ++). but decreased with increase in the concentration of electron donor impurities (Sr^{††}). The shift in the levelling out of the log (σ) curve to higher values of concentration was the result of dissociation of Pb++v+ complexes and resolution of cation vacan-

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cies in the cation	sublattice. For KCI	L-Sr ^{††} this shift di	ld not occur since	Sr ⁺⁺
	lectronic acceptor protein the process venent in the localizes.			
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IGHAT'YEVA, M. B., GOTLIB, Ye. Ye.

Horses

Development of thoroughbred riding horses up to the age of one and a half, Konevodstvo No. 4, 1952.

Monthly List of Russian Accessions, Library of Congress, July 1952. Unclassified.

"APPROVED FOR RELEASE: 04/03/2001

CIA-RDP86-00513R000518410014-7

IGNAT'YEVA, M. B.

Horse Breeding

Organized weaning of foals and proper raising during their confined period, Konevodstvo, 22, No. 3, 1952.

" Monthly List of Russian Accessions, Library of Congress November 1952 Unclassified.

CATEGORY: Farm Antiques. 101386.

A38. JOUR. FRZBiol., No. 4, 1959, No. 16626

AUTHOR: Ignot'yeve, N. 8.
INST.:
TITLE: Horse Breeding in Bungary.

ORIG. PUB.: Konyevodstvo, 1958, No 3, 37-43

ASSTRACT: No abstract.

CARO: 1/1

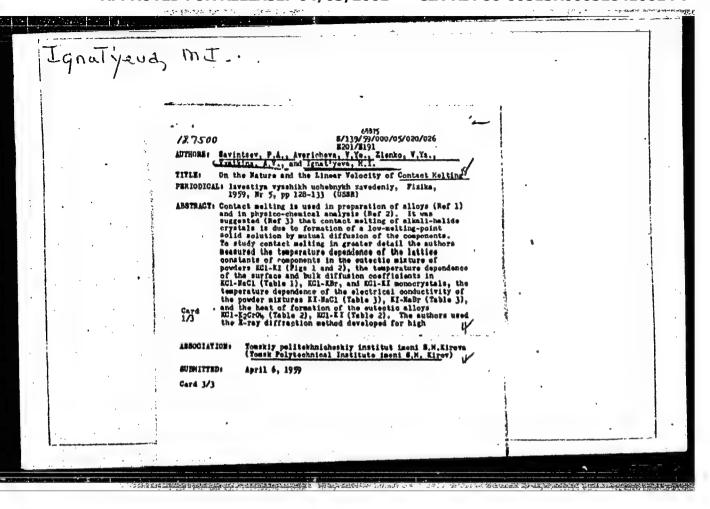
KUZ'MOV, Nikolay Terent'yevich, inzh.; ALEKSEYEV, G.P., inzh., red.;
BUSHUYEV, N.M., kand. tekhn.nauk, red.; GUTMAN, I.M., inzh., red.;
KALENICHENKO, P.T., inzh., red.; ICMAT'YEV, N.G., agronom, red.;
PICHAK, F.I., kand. tekhn.nauk, red.; POEKANOV, I.P., kand. tekhn.
nauk, red.; DUGINA, N.A., tekhn.red.

[Efficient use of machinery in harvesting by separate stages]
Ratsional noe ispel sevenie mashin na razdel noi uborke. Meskva,
Gos.nauchno-tekhn.izd-ve mashinostroit.lit-ry, 1959. 101 p.
(MRA 13:5)

(Harvesting machinery)

"APPROVED FOR RELEASE: 04/03/2001

CIA-RDP86-00513R000518410014-7



 IGNAT'YEVA, M.I.; ZAVADOVSKAYA, Yo.K.; HELIK-GAYKAZYAN, I.Ya.

Effect of divalent impurities on the radiation stability of alkali halide crystals. Fiz. tver. tela 5 no.10:2775-2779 0 (MIRA 16:11)

1. Tomskiy politekhnicheskiy institut.

ACCESSION NR: AP4028465

S/0181/64/006/004/1243/1246

AUTHORS: "Melik-Gaykazyan, I. Ya.; Zavadovskaya, Ye. K.; Ignat'yeva, M. I.

TITLE: Change in electrical conductivity of KCl crystals on addition of bivalent impurities after x-ray irradiation

SOURCE: Fizika tverdogo tela, v. 6, no. 4, 1964, 1243-1246

TOPIC TAGS: conductivity, electrical conductivity, KCl, KCl crystal, x-ray, F centor, Pb doped KCl, Sr doped KCl, F centor density, impurity, impurity concentration, current carrier, hole center

ABSTRACT: The authors have studied the ionic conductivity, its radiation change during equal doses of x-irradiation (~4.10⁴ roentgens) in KCl-Pb and KCl-Sr crystals, and the density of F centers in KCl-Sr. Pb and Sr impurities have altogether different acceptor properties relative to holes. Pb²⁺ in NaCl is an acceptor of electrons, but Sr²⁺ in KCl gives rise to activator hole centers. In KCl a comparatively small increase in electrical conductivity accompanying the injection of Sr up to 2.10⁻² molecular percent corresponds to an increase in F

ACCESSION NR: AP4028465

centers of 210%. The maximum increase in density of F centers in KCl activated by Pb does not exceed 70%. Changes in conductivity with changes in impurity concentration indicate that the first are observed only in the interval of concentration for which a change in conductivity in nonirradiated crystals takes place. Conductivity in a crystal affects radiation change only at those impurities situated in regular points of the crystal lattice. Increased radiation changes in the conductivity of KCl-Pb are observed, first, through decrease in number of current carriers arising during localization of holes at single ion vacancies and, second, because of increased stability of hole centers that have formed through the appearance of electron atomic centers. Orig. art. has: 2 figures.

ASSOCIATION: Tomskiy politekhnicheskiy institut (Tomsk Polytechnical Institute)

DATE ACQ: 27Apr64

ENCL: 00

SUB CODE:

NO REF SOV: OOL

OTHER:

STAVSKAYA, V. V., dotsent; DAVYDOVA, T. A., kand. med. nauk; IGNAT'YEVA, N. A. (Leningrad)

Clinical characteristics of an outbreak of influenza in the spring of 1961. Klin. med. 40 no.7:41-47 J1 62. (MIRA 15:7)

1. Iz kafedry propedevticheskoy terapii (sav. - deystvitel'nyy chlen AMN SSSR prof. M. D. Tushinskiy[deceased]). I Leningrad-skogo instituta imeni akad. I. P. Pavlova)

(INFLUENZA)

江 — 在在公司的全部制备各种

YERMOSHERKO, M.A.; IGNAT'YEVA, N.P.

Methods for irrigating cotton in growing containers. Dokl.AN Uz.SSR no.11:45-48 '56. (MIRA 13:6)

1. Institut sel'skogo khozyaystva AN UESSR. Predstavleno chlenomkorrespondentom AN UESSR A.I.Abtonomovym. (Cotton growing)

1、2012年12年 大学教师 安全的报报报程等活动第二十二年 (1) · · · ·

BELYAYKINA, I.V., inzh.; IGNAT'YEVA, N.G., inzh.

Nomographs for calculating the strength of welded heating system
pipes. Elek.sta. 32 no.6:23-26 Je '61. (MIRA 14:8)

(Steam pipes) (Heating from central stations)

生物的影響等的作為英國企業等。1981年1991年,

PESHKOVA, V.M.; IGNAT'YEVA, N.G.

1,2-Cycloheptanedione dioxime as a reagent for the gravimetric and extraction-photometric determination of nickel in the presence of copper. Zhur.anal.khim. 17 no.9:1086-1090 D 162.

(MIRA 16:2)

1. M.V. Lomonosov Moscow State University.
(Nickel—Analysis) (Cycloheptanedione)

PESHKOVA, V.M.; IGNAT YEVA, N.G.; OZEROVA, G.P.

Determination of rhenium with X-furyl dioxime in the presence of molybdenum. Zhur.anal.khim. 18 no.41496-499 Ap 163.

(MIRA 1616)

1. M.V.Lomonosov Moscow State University.
(Rhenium—Analysis) (Molybdenum—Analysis)

YAN TOUL; IGNAT'YEVA, N.G.; PESHKOVA, V.M.

Valency of rhenium during its reduction. Zhur. anal. khim. 19 no.2:224-228 64. (MIRA 17:9)

1. Moskovskiy gosudarstvennyy universitet imeni Lomonosova.

PEGHKOVA, V.M.; IGNAT'YEVA, N.G.

Complex formation of molybdenum with some dioximes. Zhur.anal.khim. 19 no.10:1269-1270 *64. (MIRA 17:12)

1. M.V.Lomonosov Moscow State University.

"APPROVED FOR RELEASE: 04/03/2001 CIA-RDP86-00513R000518410014-7

L 23622-65 EWT(m)/EWP(t)/EWP(b) IJP(c) JD/JG/HLK

ACCESSION NR: AT5002789 S/0000/64/000/000/0239/0241

AUTHOR: Ignat'yeva, N. G.; Peshkova, V. M.

TITLE: Determination of rhenium in the presence of molybdenum, tungsten, and vandium

SOURCE: Vsesoyuznoye soveshchaniye po probleme reniya. 2d, Moscow, 1962. Reniy (Rhenium); trudy soveshchaniya. Moscow, Tzd-vo Nauka, 1964, 239-241

TOPIC TAGS: rhenium determination, rhenium analysis, spectrophotometry,

ABSTRAGT: The authors determined rhenium in the presence of large amounts of molybdenum Re:Mo ratios were 1:40 and 1:100) by means of a differential spectro-valence state of Re (V or IV) and promotes a faster formation of the compound between rhenium and &-furyldioxime. The simplification introduced by the authors consisted in taking as the blank a definite amount of the solution being analyzed, determination of rhenium. Using this simplified method, the authors also Card 1/2

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1:1000). Finally, rhenium was determined in the presence of a 10,000-fold excess of vanadium, which does not interfere with the determination, by means of a direct spectrophotometric analysis. Orig. art. has: 1 figure, 3 tables

ASSOCIATION: None

SUMMITTED: 05Aug64

ENCL: 00

SUB CODE: IC,GC

NO REF SOV: 001

OTHER: 002

Cord 2/2

26285 S/078/61/006/009/005/010 B107/B101

18.8300 AUTHORS:

Kochergin, V. P., Ignat'yeva, N. I.

TITLE:

Oxidation of iron in melts containing sodium halogenides and

sodium carbonate

PERIODICAL: Zhurnal neorganicheskoy khimii, v. 6, no. 9, 1961, 2126 - 2131

TEXT: The rate of oxidation of Armco iron in mixtures of sodium carbonate with NaF, NaCl, NaBr and NaI between 700 and 900°C was investigated. The degree of thermal dissociation of Na₂CO₃ in such melts at 800°C and the emf of a galvanic cell iron - melt - platinum were also determined. The investigation of the rate of oxidation of iron is of interest in order to clarify the nature of the adhesive forces between enamels and the metallic surface. Fig. 1 shows the change of the rate of oxidation at 700°C in Na₂CO₃. NaX (X = F, Cl, Br, I) melts with 50 mole% Na₂CO₃. The aggressiveness drops in the order NaI, NaBr, NaF, NaCl. This is based on the differently strong depassivating effect of the halide ions. It was roentgenographically established that wistite and magnetite form as reaction products in melts with Card 1/6

26285 3/078/61/006/009/005/010 B107/B101

Oxidation of iron in melts containing ...

NaF and NaCl, and wüstite alone in melts with NaBr and NaI. The oxidation proceeds according to the equation Fe + CO_2 = FeO (or Fe₃O₄) + CO (1). The oxidation products form a coat on the iron which has, however, a porous structure and does not prevent further oxidation. Only a small part of the iron dissolves as sodium ferrite. The degree of dissociation of Na₂CO_x in the melts of the composition mentioned was determined at 800°C (Fig. 2). Here, too, the order NaBr, NaF, NaCl corresponds to a decreasing degree of dissociation. No trivalent iron forms in the melts with NaBr and NaI during oxidation of the iron, probably because the Fe_{504}^{0} from E_{q} (1) is reduced by the halide to FeO. The Br or I thus formed has a strong oxidizing effect on the iron; the more aggressive effect of the bromide and iodide, especially with access of air, is explained in this way (Fig. 1, isotherm 1). The rate of oxidation in melts with various halids concentration (8000C, 1 hr) was also investigated. As shown in Fig. 4, there is a strong concentration dependence, i.e., maximum aggressiveness exists for certain concentrations. The emf of the galvanic cell iron - melt - platinum at 800°C was finally determined. The melt consisted of $Na_2CO_3 - NaX (X = F, Cl, Br)$ in the molar

Card 2/6

26285 S/078/61/006/009/005/010 B107/B101

Oxidation of iron in melts containing ...

ratio 1: 1. Fig. 6 shows the change of the emf with time. It is stated in conclusion that a Na₂CO₃ - NaCl melt with 30 - 50% NaCl is least aggressive.

G. V. Akimov (Osnovy ucheniya o korrozii i zashchite metallov, Metallurgizdat, 1941); N. D. Tomashov, V. I. Modestova (Tr. In-ta fiz. khimii AN SSSR, 5, 75 (1958)); B. N. Kabanov et al. (Dokl. AN SSSR, 59, 917 (1948); Zh. fiz. khimii, 31, 2501 (1957)); Z. A. Ioffa (Zh. fiz. khimii, 13, 1105 (1939)) and O. A. Yesin et al. (Fizicheskaya khimiya pirometallurgicheskikh protsessov, Metallurgizdat, 1950) are mentioned. There are 6 figures and 23 references: 19 Soviet and 7 non-Soviet. The four most recent references to English-language publications read as follows: O. Balestra. Metall Progress, 1, 1957; F. Bacon, I. Forrest. The Engineer, 202, 93 (1956); F. Bacon. J. Beama, 61, 6 (1954); M. E. Straumanis, A. W. Schlechten, J. Electrochem. Soc., 102, 131 (1955).

ASSOCIATION: Ural'skiy gosudarstvennyy universitet im. A. M. Gor'kogo (Ural State University imeni A. M. Gor'kiy)

SUBMITTED: July 19, 1960

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Card 3/6

DAVYDOV, V.I.; BELIKOV, A.M.; IGNAT YEVA, N.I.; VERBOVETSKAYA, D.Ye.

Reaction of germanium dioxide with iron. Zhur.prikl.khim. 35 no.11: 2543-2546 N *62. (MIRA 15:12) (Germanium oxide)

Country

USSR

Cartoffory

: CULTIVATED PLANTS. MEDICINAL. Essential Oils. Toxins.

f.bs. Jour. REFZHUR-BIOL., 21,1958, NO-96182

Author

: Ignat'yeva . N.S.

Institut.

: Moscow Pharmacoutical Institute

Midtle.

: A Pharmacognostic Study of the Tansy

Orig. Pab. : Sb. nauchn. racot. Hosk. farmatsevt. in-t, 1957,

1, 187-200

Abstract

: The tanay (Tanacetum vulgare, d) is a perennial herb which has found widespread use in folk medicine since time immemorial. A exact botanicoanatomical study is made of the leaves, stems, roots, rootstocks, inflorescences and fruit. observations were conducted over a period of two years (1954-1955) throughout the entire vegetative stage. A detailed description is presented of the external and inner structure of the inflorescences (standard raw material) with an indica-

Card:

1/2

"APPROVED FOR RELEASE: 04/03/2001 CIA-RDP86-00513R000518410014-7

М · USSR Country Category CULTIVATED PLANTS, MEDICINAL. Essential Oils. Toxins. Abs. Jour. : REF ZHUR-BIOL., 21,1998, NO-96183 Authorn : Ignat'yeva, N.S. : Moscow Pharmaceutical Institute in thit tot. 5.41a : A Pharmacognostic Study of the Tansy. II. Phytochemical Research. ortg. Prb. :Sb. nauchn. rabot. Mosk. farmatsett. in-t. 1957. 1. 201-208 During 1954-1955 a study was made of the dynamics Abstract of the accumulation of essential oil (I) and tanning (II) in the leaves and blossoms of the tansy (III) throughout the entire vegetation period. The content of I and II reached its maximum during the stage from the beginning of budding to the end of flowering with a decrease in the fruiting stage. The richest I and II was in the flowers and III in the leaves. The atem and root system practically contained no I, and their II content was lower than in the flowers and leaves. I, derived from Card: 1/3

APPROVED FOR RELEASE: 04/03/2001 CIA-RDP86-00513R000518410014-7"

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"APPROVED FOR RELEASE: 04/03/2001 CIA-RDP86-00513R000518410014-7

Country Catuzory CULTIVATED PLANTS. MEDICINAL Abs. Jour. : REF ZHUR-BIOL., 21,1958, NO-9618 3 Author Institut. : Titlo Orig. Pub. : Abstract : from the flowers in full blossom had more esters and thujone than I at the start of flowering. I from the leaves had in all plant development stages studied nearly identical essential oil content, although before budding and full flowering more thujone was seen. High polyphenol centent in I in the flowers (47.87%) and in III in leaves (33.94%) were noted. It is suggested the raw material be collected (flowers and leaves) in the budding, start of flowering and full flowering stages. Card: 2/3

IGNAT'YEVA, N.S.

Analysis of Tanacetum vulgare for the presence of aromatic oxacids. Apt. delo 9 no. 4:26-28 Jl-Ag 160. (MIRA 13:8)

l. Kafedra farmakognozii (sav. - prof. L.A. Pazdorskaya) farmatsevticheskogo fakuliteta I Moskovskogo ordera Lerina meditainskogo instituta im. I.M. Schenova. (TANSY)

(MIRA 13:10)

IGNAT'YEVA, N.S. Anatomic structure of Tanacetum vulgare L. Apt. delo 9 no. 5:25-29 S-0 160.

1. Kafedra farmakognozii (nauchnyy rukovoditel! - prof. L.A. Razdorskay) I Moskovskogo ordena Lenina meditsinskogo instituta Razdorskay, I rocking im. I.M. Sechenova. (TANSY)

IGNAT'YEVA, N. S.

Cand Pharm Sci - (diss) "Accumulation of active substances in the common tansy grown in the Moscow Oblast, and its pharms-cognostic evaluation." Moscow, 1961. 19 pp; (Ministry of Public Health RSFSR, First Moscow Order of Lenin Medical Inst imeni I. M.Sechenov); 250 copies; price not given; (KL, 6-61 sup, 242)

IGNAT'YEVA, N.S.

Analysis of Tanacetum vulgare for the presence of manganese and the influence of manganese salts on the accumulation of ethereal oil and tanning substances. Apt. delo 10 no.3:19-24 My-Je '61.

(MIRA 14:7)

1. Kafedra farmakognozii farmatsevticheskogo fakul'tota (rukovodital'prof. L.A.Razdorskaya [deceased]) I Moskovskogo ordena Lenina
meditsinskogo instituta imeni I.M.Sechenova.

(TANSY)

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GRINKEVICH, N.I.; IGNAT'YEVA, N.S.; L'VOVA, I.L.; ZORIN, Ye.A.

Examination of some vitamin-containing plants for their manganese content. Apt. delo. 11 no.5:41-43 S-0 '62.

(MIRA 17:5)

1. Farmatsevticheskiy fakul'tet I Moskovskogo ordena Lenina meditsinskogo instituta imeni Sechenova.

DOLGOVA, A.A.; IGNAT'YEVA, N.S.

Morphological and anatomical characteristics of oleander Morphological and anatomical communications leaves. Apt. delo 12 no.4:36-41 J1-Ag '63. (MIRA 17:2)

1. Farmatsevticheskiy fakulitet, 1-go Moskovskogo ordena Lenina meditsinskogo instituta imeni I.M. Sechenova.

GRINKEVICH, N.I.; IGNAT LEVA, N.S.; SAFRONICH, L.N.

Examination of some representatives of the Compositae family for manganese and carotens content. Apt. delo 12 no.2:38-40 Mr-Ap (63. (MIRA 17:7)

1. Farmatsevticheskiy fakulitet I Moakovskogo ordena Lenina meditsinskogo instituta imeni I.M. Sechenova.

 GRINER, B.M.; GRINKEVICH, N.I.; IGNAT'YEVA, N.S.; KAZ'MINA, L.P.

Color of leaves as an index of the content of tanning substances in plants. Biul. Glav. bot. sada no.53:72-75
164. (MIRA 17:6)

1. Botanicheskiy sad Pervogo moskovskogo meditsinskogo instituta imeni Sechenova.

HELOV, N.S.; BIRYUKOV, I.V.; VERBLYUDOV, N.N.; GORBUNOVA, M.N.; YESIPOVA, M.M.;

IL'ICHEV, A.I.; IGNAT'YEVA, H.Ya.; KOVACHEVICH, P.M.; LYTKIN, A.M.;

LOSKUTOV, V.G.; MAZYUKOV, A.S.; MIROSHNICHENKO, N.Ya.; NEFEDOV, A.Ya.;

OSIPOV, K.V.; OSIPOV, P.W.; PETROV, N.G.; PETRACHKOV, M.I.;

PINEVICH, K.M.; POPOV, B.B.; POTAPOV, P.V.; PREDEIN, F.Ye.; PUKHOV, A.F.;

CHUSOVITINA, Ye.I.; ANGEL'SKIY, W., tekhn.red.

[The Kuznetsk Basin in the sixth five-year plan] Kuxbass v shestoi piatiletke. [Kemerovo] Kemerovskoe knizhnoe izd-vo, 1956. 125 p.
(MIRA 10:12)

(Kusnetsk Basin)

IGNATYEVA, O.A., ZAIKONNIKOVA, I.V., AGONSKAYA, L.S.

Antibacterial properties of % organic compounds of phosphorus.

Khimiya i Primeneniye Fosfororganicheskikh Soyedineniy (Chemistry and application of organophosphorus connounds) A. TE. AREUZOV, Ed. Publ. by Kazan Affil. Acad. Sci. USSR, Moscow 1962, 632 pp.

Collection of complete papers presented at the 1959 Kazan Conference on Chemistry of Grganophosphorus Compounds.

PCZDEYEV, K.A., starshiy nauchnyy sotrudnik; IGNAT'YEVA, C.A., mladshiy nauchnyy sotrudnik

Use of the method of indirect hemagglutination reaction in the diagnosis of brucellosis. Uch. zap. KVI 89:75-78 162. (MIRA 18:8)

1. Laboratoriya Nr. 2 (zav. - prof. Kh.C.Gizatullin) Kazanskogo vaterinarnogo instituta.

VELIKORETSKIY, D.A.; LORIYE, K.M.; FINKEL', I.I.; GRIGORCHUK, Yu.F.;

BERGER, L.Kh.; "UTROBINA, V.V.; KHARCHENKO, V.P.; MESHCHERYKOV, A.V.,

student V kursa; OBEREHCHENKO, Ya.V., kand.med.nauk; NIKITIN, A.V.;

MUKHOYEDOVA, S.N.; KUSMARTSEVA, L.V., assistent; KUZNETSOV, V.A.,

dotsent; KUKHTINOVA, R.A., assistent; BOHDARENKO, Ya.D. (g. Fastov);

KURTASOVA, L.V. (g. Fastov); PEVCHIKH, V.V.; CHURAKOVA, A.Ye.;

BABICH, M.M.; KUZ'MIN, K.P.; PAVLOV, S.S.; SHEVLYAKOV, L.V., kand.

med.nauk; IGHAT'YEVA, O.M.; ZEYGERMAKHER, G.A.; GUTKIN, A.A.;

POLYKOVSKIY, T.S.

Resumes. Sov.med. 25 no.11:147-152 N 161.

(MIRA 15:5)

1. Iz Instituta grudnoy khirurgii AMN SSSR (for Velikoretskiy, Loriye, Finkel'). 2. Iz bol'nitsy No.3 Gorlovki Stalinskoy oblasti (for Grigorchuk). 3. Iz Tyumenskoy oblastnoy bol'nitsy (for Berger, Utrobina). 4. Iz Karatasskoy rayonnoy bol'nitsy Yuzhno-Kazakhstanskoy oblasti (for Kharchenko). 5. Iz Gospital'noy khirurgicheskoy kliniki I Moskovskogo ordena Lenina meditsinskogo instituta imeni Sechenova (for Meshcheryakov). 6. Iz kliniki propedevticheskoy terapii Stalinskogo meditsinskogo instituta na baze oblastnoy klinicheskoy bol'nitsy imeni Kalinina (for Oberemchenko). 7. Iz kliniki gospital'noy terapii Voronezhskogo meditsinskogo instituta (for Nikitin, Mukhoyedova).
8. Iz kafedry obshchey khirurgii Kishinveskogo meditsinskogo instituta (for Kusmartseva). (Continued on next card)

VELIKORETSKIY, D.A. -- (continued) Card 2.

9. Iz akushersko-ginekologicheskoy kliniki Stalinskogo meditsinskogo instituta na baze bol'nitsy imeni Kalinina (for Kuznetsov, Kukhtinova).
10. Iz gospital'noy terapevticheskoy kliniki Izhevskogo meditsinskogo instituta (for Pevchikh, Churakova). 11. Iz Nosovskoy rayonnoy bol'nitsy Chernigovskoy oblasti (for Babich). 12. Iz Vyborgskoy mezhrayonnoy bol'nitsy (for Pavlov). 13. Iz 1-y gorodskoy bol'nitsy Tyumoni (for Ignat'yeva). 14. Iz 2-y infektsionnoy bol'nitsy g. Zaporozh'ya (for Zeygermakher). 15. Iz infektsionnogo i prozektorskogo otdeleniy Petrozavodskoy gorodskoy bol'nitsy (for Gutkin, Polykovskiy).

(MEDICINE—ABSTRACTS)

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AUTHOR: Ignatiyeva, O. V.

TITLE: Dynamic characteristics of the gas pipe system of a marine engine and a turbosupercharger

SOURCE: Ref. zh. Vodnyy transport, Abs. 6V78

REF SOURCE: Tr. Tsentr. n.-i. in-ta morsk. flota, vyp. 63, 1965, 73-80

TOPIC TAGS: turbosupercharger, ship, marine engine, gas dynamics

ABSTRACT: Evaluations were made of the main functions relative to the gas dynamics of turbosuperchargers, taking the receiver volumes into account. It was found that in the evaluation of its dynamic characteristics during intermittent disturbance, the turbosupercharger can be considered an aperiodic link of the first order. Since the magnitude of the time constant of the engine as the controlling object of the rotation speed of the shaft is of the range of tenths of a second, the turbosupercharger exerts a sufficient effect on the transitory process. [Translation of abstract]

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UDC: 621. 431, 74-501, 22+621, 515, 5-501, 22

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m)/WT(m)/FWP(f)/EVA(d)/T-2/EVA(1) AT6008033 SOURCE CODE: UR/2752/65/000/063/0073/0080 AUTHOR: Ignat'yeva, O. V. ORG: none TITLE: Dynamic characteristics of the gas channel of a marine engine and turbine supercharger SOURCE: Leningrad. Tsentral nyy nauchno-issledovatel skiv institut morskogo flota. Trudy, no. 63, 1965. Tekhnicheskaya ekspluatatsiya morskogo flota (Technical operation of the merchant marine), 73-80 TOPIC TAGS: marine engine, diesel engine, supercharged engine, engine performance characteristic, gas flow dynamics supercharger ABSTRACT: The air-gas channel of a marine engine with a gas-turbine supercharger can be regarded as a complex branched annular duct with artificial gas-flow turbulization. The working processes are considered separately for the engine, the supercharger, the receiver, and the condenser of scavenging air. The similarity theory is widely used for analyzing varying operating conditions. As graphically represented, the calculated characteristics of various superchargers show a linear relationship over a wide range of analyzed parameters. Thus, the analyzed linearized equations given for the dynamic characteristics of the supercharger and receiver are applicable for small as well as extensive disturbing effects; this was proved by reference to test data on an 18,000-hp UDC: 621.431.74-501.22+621.515.5-501.22 CONTRACTOR OF THE PERSON OF TH

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diesel engine installed on the tanker Lisichansk and by sudden load changes occurring when the rpm of the fuel pumps were varied from 90 to 114. Thus, as shown, the theoretically plotted curves for the air-gas-channel dynamics adequately represent the transition stages during sudden load changes even in the presence of high disturbing effects. Orig. art. has: 4 figures and 7 formulas.											
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ACCESSION NR: AR5019473

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SOURCE: Ref. zh. Dvigateli vnuirennego sgoraniya, Otdel'nyy vypusk, Abs. 7.39.211

AUTHOR: Antonovich, S.A.; Ignat'yeva, O.V.

TITLE: Dynamic properties of diesel units

CITED SOURCE: Tr. Tsentr. n.-i. in-ta morsk. flota, vyp. 59, 1964, 14-36

TOPIC TAGS: engine control system, diesel engine, marine engine, turboshaft engine, supercharged engine, shaft 1/1/

TRANSLATION: The authors discuss the dynamic properties of a marine diesel as a system controlling the rpm of a shaft in marine diesel and diesel-generator installations with and without a gas turboblower. The analysis covers smooth and rough water operations of engines with a turboblower and an ideal or dynamically complex regulator of shaft rpm. Finally, authors describe ways of improving the static and dynamic properties of controlled objects, so as to insure optimal characteristics of the transient process.

SUB CODE: IE, PR

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IGNAT'YEVA, O.V.

Dynamic characteristics of the gas duct of a marine engine and turbosupercharger. Trudy TSNIIMF no.63:73-80 (MIRA 18:12)